

This article was downloaded by: [Tomsk State University of Control Systems and Radio]

On: 23 February 2013, At: 05:49

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954

Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/gmcl16>

Mesomorphic Derivatives Of Ferrocene

Jacques Malthete^a & Jean Billard^a

^a Laboratoires de Chimie Organique des Hormones et de Physique de la Matière Condensée COLLEGE de FRANCE, 75231, Paris Cedex 05

Version of record first published: 28 Mar 2007.

To cite this article: Jacques Malthete & Jean Billard (1976): Mesomorphic Derivatives Of Ferrocene, *Molecular Crystals and Liquid Crystals*, 34:5, 117-121

To link to this article: <http://dx.doi.org/10.1080/15421407608083898>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.tandfonline.com/page/terms-and-conditions>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages

whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

MESOMORPHIC DERIVATIVES OF FERROCENE

by Jacques MALTHETE and Jean BILLARD

Laboratoires de Chimie Organique des Hormones et de Physique de la Matière Condensée,
COLLEGE de FRANCE, 75231 Paris Cedex 05.

Abstract The first series of ferrocene derivatives exhibiting mesomorphic properties (substituted N-(p-benzoyloxybenzylidene) anilines 1a and 1b) has been synthesized for application in Mössbauer spectroscopy.

(Received: November 1, 1976)

During the past ten years, Mössbauer spectroscopy has become a very useful technique for the study of solid state by means of heavy metal nuclei such as ^{57}Fe , ^{61}Ni , ^{67}Zn , ^{119}Sn ... Lately, its utilization in structural studies of ordered systems such as smectic liquid crystals has undergone remarkable development (1).

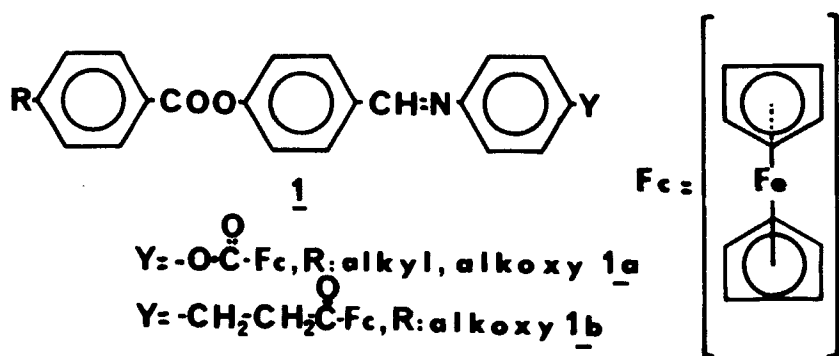
Hitherto, only non-mesomorphic "Mössbauer markers" have been used, for example : diacetyl-ferrocene (2) and tin-containing compounds (3) (4) dissolved in smectic phases. The principal limitation of these "markers" appears to be absence of mesophases and low solubility.

Besides, by such a method, the results concerning the preponderating constituent can be

only indirect and their interpretation is often difficult *.

Consequently, we felt that it was highly desirable to undertake the preparation of stable and easily synthesized liquid crystals, incorporating appropriate Mössbauer metal atoms and having relatively low melting points.

Ferrocenes appeared to correspond the best to the requirements: they are chemically stable, and readily synthesized. Yet, it was quite difficult to conceive a mesogenic organic structure containing a voluminous group such as ferrocene, perpendicular to the molecular axis which would yield the desired mesophases (6).



* To our knowledge, only one mesomorphic compound, suitable for Mössbauer experiments, has been reported to date. This substance, prepared by YOUNG (5) is a Schiff base with trimethyltin group which exhibited two smectic phases in the 165-173° C range.

On the basis of other studies, we were led to consider Schiff bases of type 1, which surprisingly tolerate large Y groups without losing their mesomorphic properties. So, we have synthesized a new series of mesomorphic derivatives of 1, containing a ferrocenyl group which have the desired properties. The melting points and transition temperatures of these compounds, determined with a differential scanning calorimeter (Perkin-Elmer DSC-2), are given in Table I ([N-4(4'-alkyl, alkoxybenzoyloxy)benzylidene]p-aminophenyl ferrocenecarboxylates 1a), and Table II ([N-4(4'-alkoxybenzoyloxy)benzylidene]1-ferrocenyl-3-(p-aminophenyl)-1-propanones 1b).

The mesophases have been identified by the examination of their texture under the polarizing microscope (Panphot Leitz) with heating stage. The textures of the nematic phases are "schlieren" with threads.

Among these new compounds, only p-alkoxy 1a derivatives exhibit stable nematic phases.

The synthesis of the Schiff bases was carried out by reaction of the corresponding p-benzoyloxybenzaldehydes 2 (0,1 mmole) prepared according to ref. (7) with p-aminophenylferrocenecarboxylate 3b (0,1 mmole-m.p.149-150°C, ethanol) or 1-ferrocenyl-3-(p-aminophenyl)-1 propanone 4b (0,1 mmole-m.p. 128-129°C, benzene/cyclohexane 1/1) in absolute ethanol at reflux (70-80 % - orange crystals - after recrystallization in absolute ethanol).

3b and 4b were obtained by catalytic reduction (H_2 , Pd-C 5 %, ethanol) of ester 3a and

and chalcone 4a respectively prepared according to ref. (8) (34 %; m.p. 140°C, ethanol) and according to ref. (9) (65 %; m.p. 195°C, methoxy-ethanol).

The study of other mesomorphic derivatives of 1 is in progress.

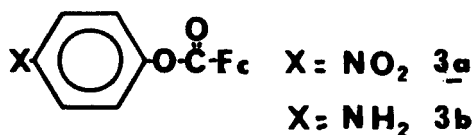
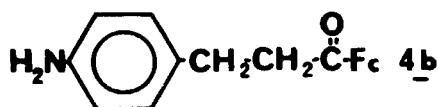
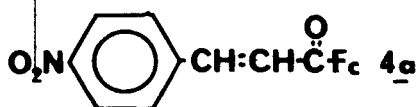
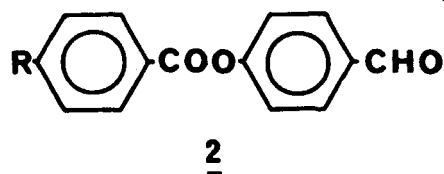
We thank Professors P.G. De GENNES and J. JACQUES for helpful discussions.

TABLE I Transition temperatures of alkyl, alkoxy 1a

	C	N	I
nC_5H_{11}	182	[155]	
nC_6H_{13}	164	[143]	
nC_8H_{17}	152	[135]	
$nC_8H_{17}O$	153	167	
$nC_{10}H_{21}O$	143	159	

TABLE II Transition temperatures of alkoxy 1b

	C	N	I
$nC_8H_{17}O$	137	[112]	
$nC_{10}H_{21}O$	135	[107]	



REFERENCES

- (1) P.G. De Gennes, J. Phys. 36, 603 (1975).
- (2) J.M. Wilson, D.L. Uhrich, Mol. cryst. liq. cryst. 25, 113 (1974).
- (3) D.L. Uhrich, V.O. Aimiwu, P.I. Ktorides, W.J. Laprice, Phys. Rev. A 12, 211 (1975).
- (4) D.L. Uhrich, Y.Y. Hsu, D.L. Fishel, J.M. Wilson, Mol. cryst. liq. cryst. 20, 349 (1973).
- (5) W.R. Young, I. Haller, D.C. Green, Mol. cryst. liq. cryst. 13, 305 (1971).
- (6) L. Verbit, T.R. Halbert, Mol. cryst. liq. cryst. 30, 209 (1975).
- (7) J.S. Dave, G. Kurian, J. Phys. 36 C₁ 403 (1975).
- (8) N. Bagget, A.B. Foster, A.H. Haines, M. Stacey, J. chem. Soc. 3528 (1960).
- (9) J. Boichard, J.P. Monin, J. Tirouflet, Bull. Soc. Chim. 851 (1963).